

Supplementary Information

Reductive Degradation of Perfluoroalkyl Compounds with Aquated Electrons Generated from Iodide Photolysis at 254 nm

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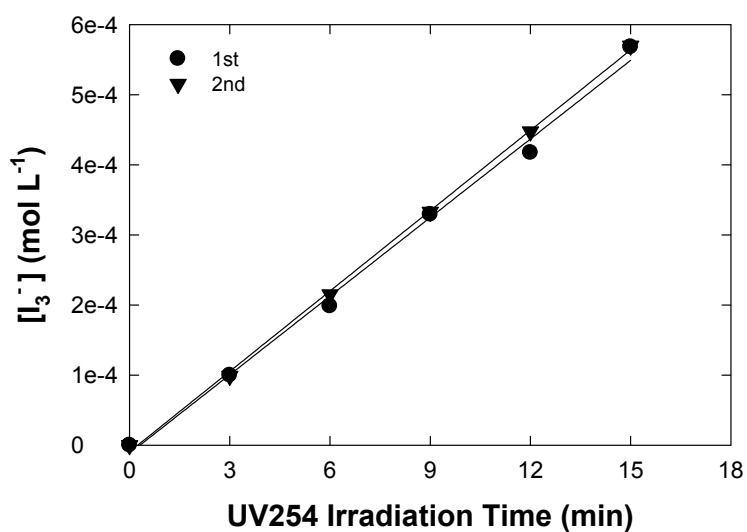


Fig. S1. Chemical actinometry determined for UV 254 nm using aqueous iodide and iodate. Incident Photon Number = $5.17(\pm 0.06) \times 10^{-5}$ mol/L/min. The intrinsic quantum yield (73%) was reflected.

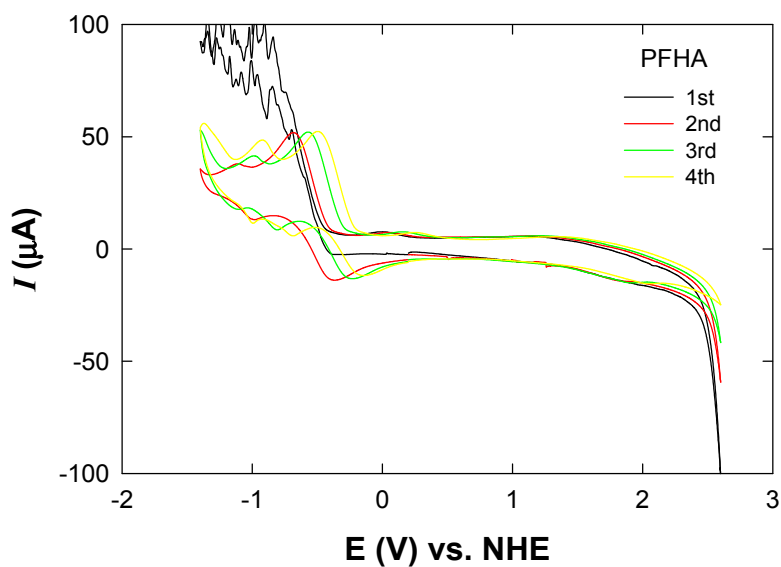


Fig. S2. Cyclic voltammograms of PFHA.

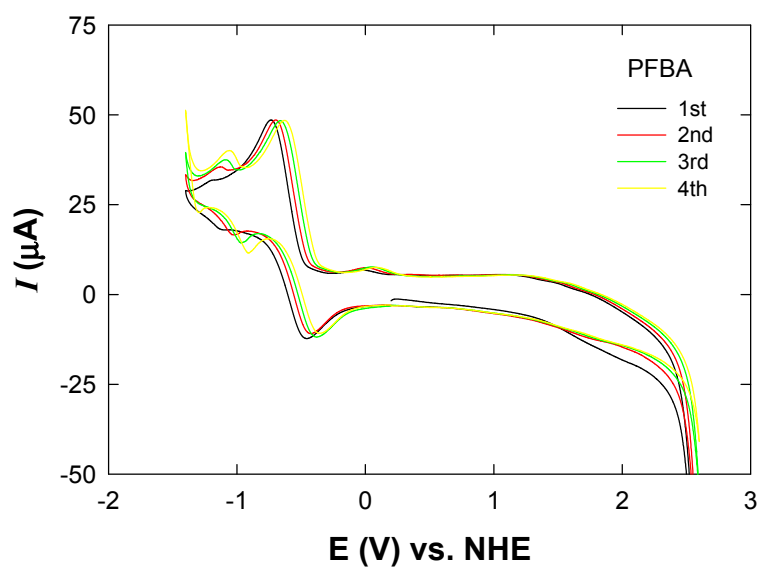


Fig. S3. Cyclic voltammograms of PFBA.

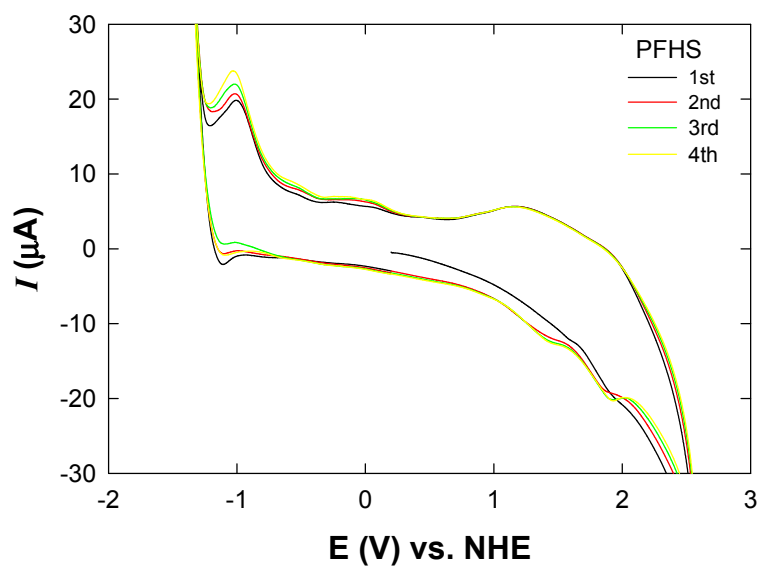


Fig. S4. Cyclic voltammograms of PFHS.

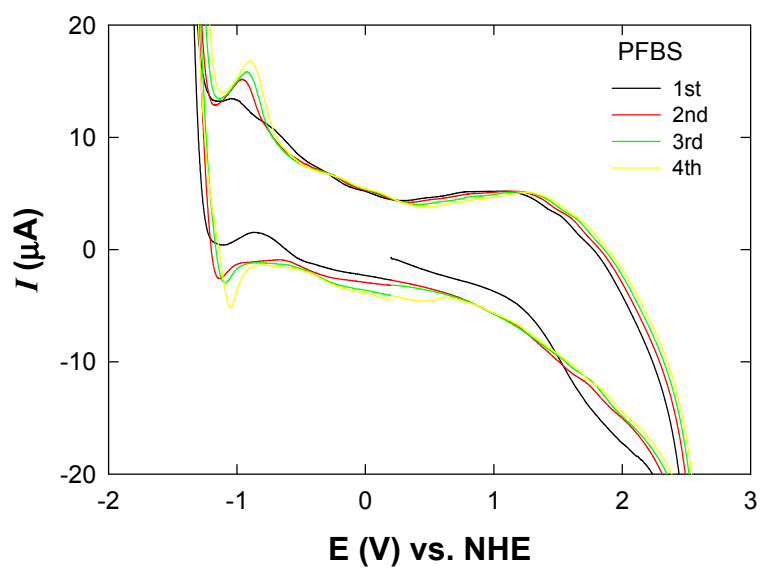


Fig. S5. Cyclic voltammograms of PFBS.

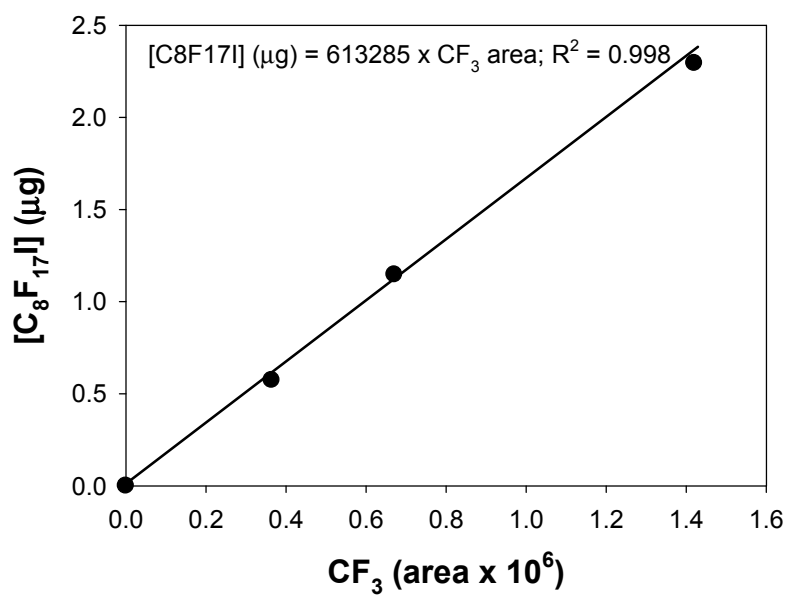
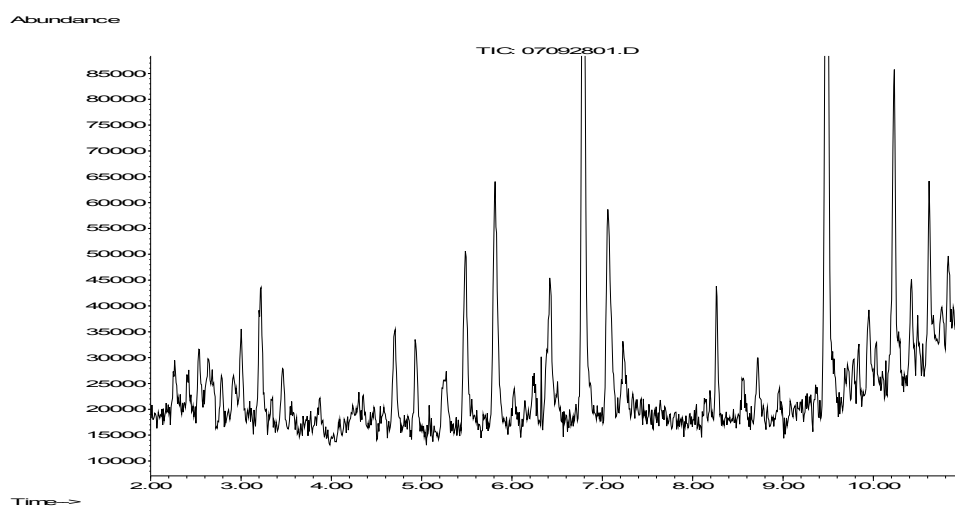
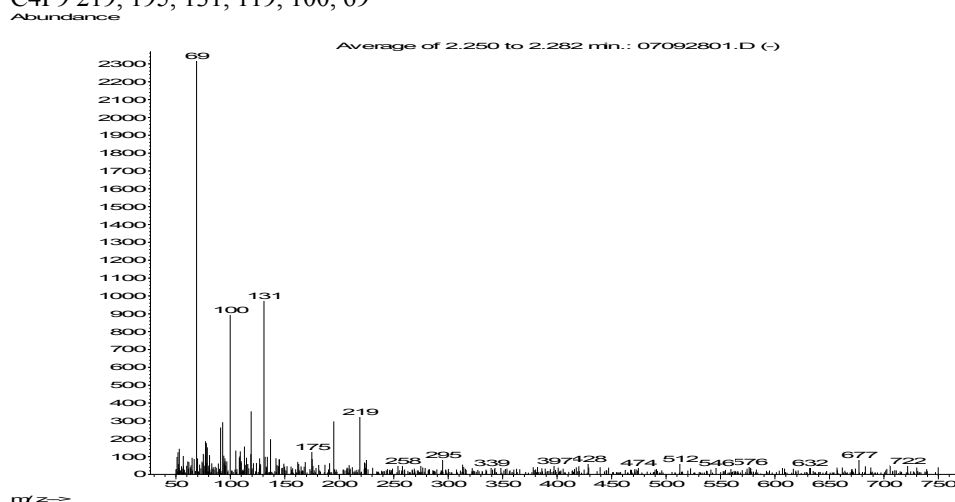


Fig. S6. Correlation between perfluorooctyl iodide concentration and its CF_3 area.

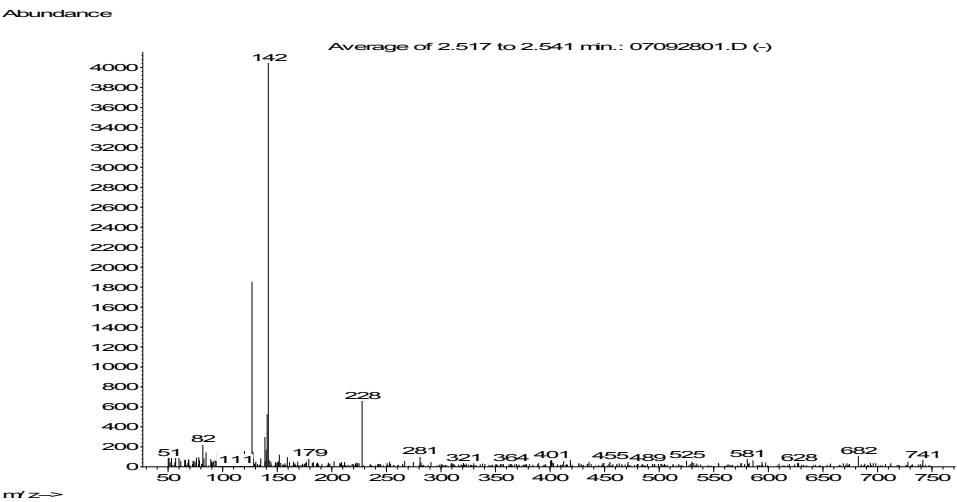
Fig. S7 (below). Total ion chromatogram (TIC) of gaseous intermediates trapped during the course of 2.5h-photolysis of aqueous PFOS with KI of 10 mM, and extracted ion chromatograms (EIC).



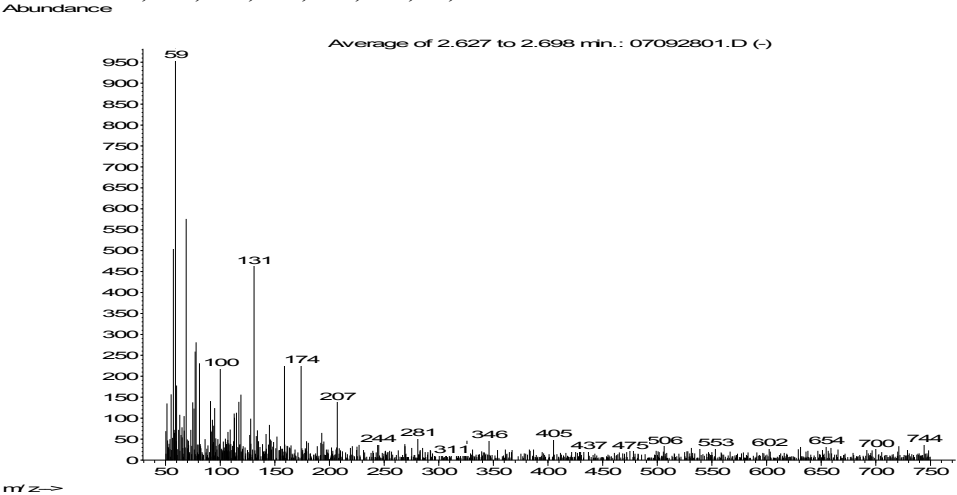
C4F9 219, 195, 131, 119, 100, 69



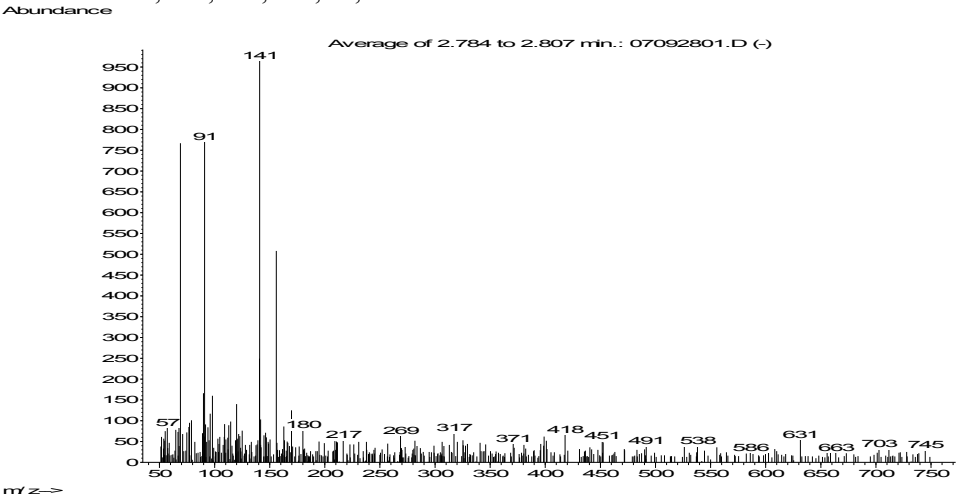
C2F4HI 228, 142, 127, 69



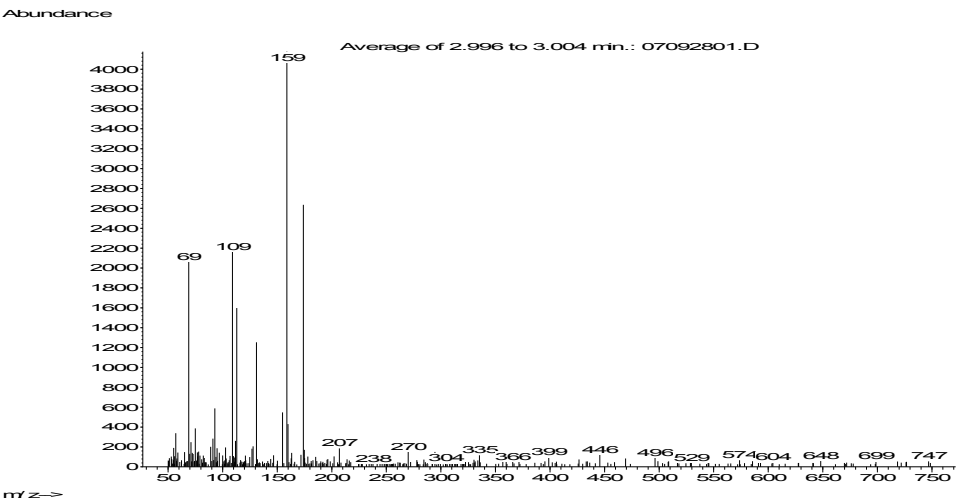
C7F6H9 207, 174, 159, 131, 119, 100, 78, 59



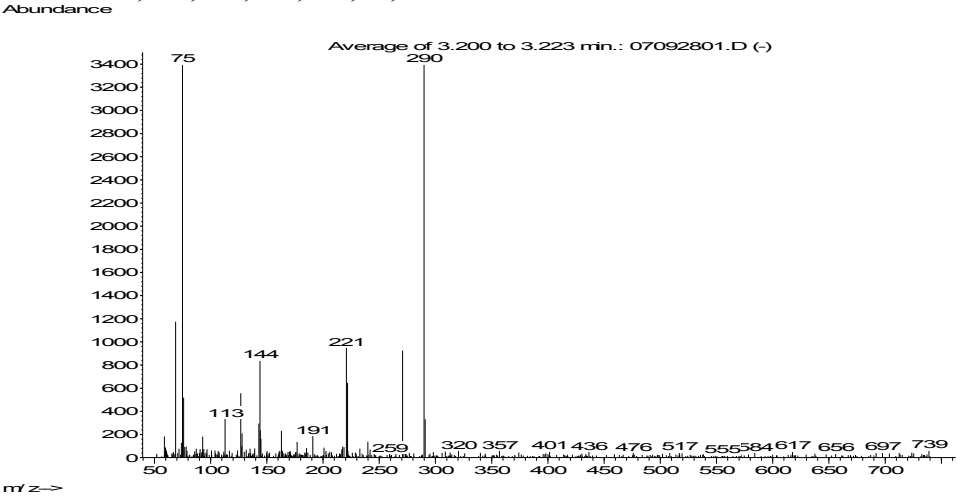
C7F4H10 170, 156, 141, 120, 91, 69



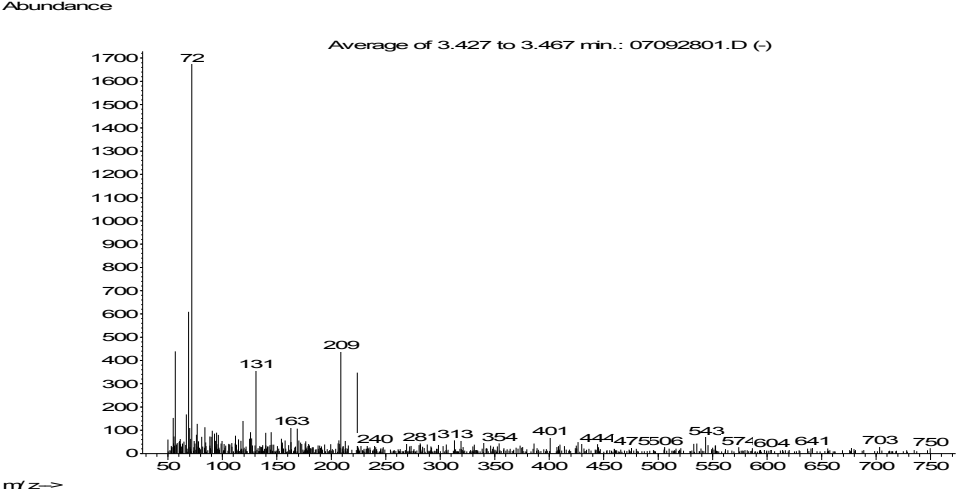
C6F5H7 174, 159, 131, 113, 109, 93, 75, 69, 57



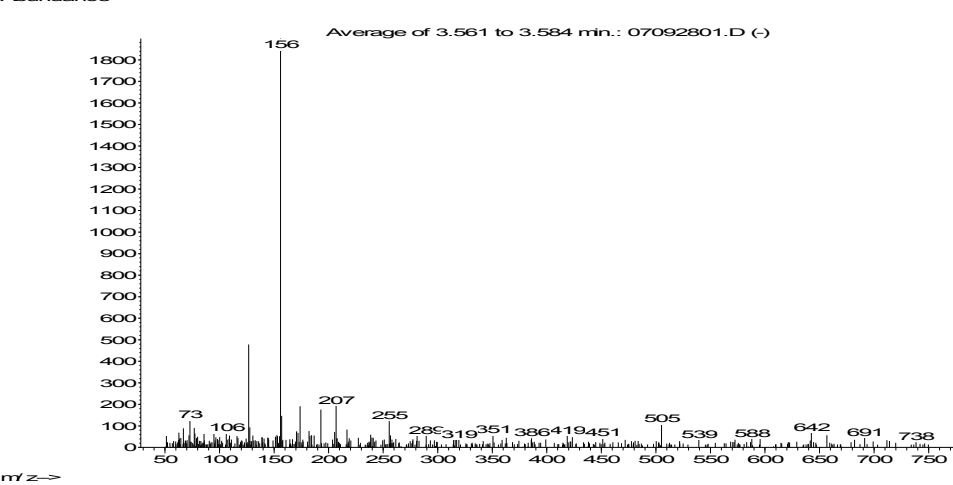
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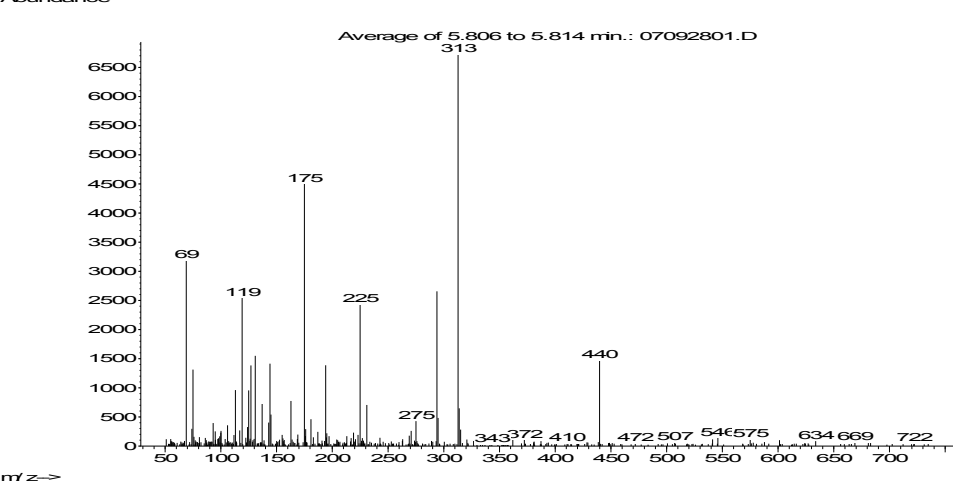
C7F7H7 224, 209, 131, 72, 69, 57



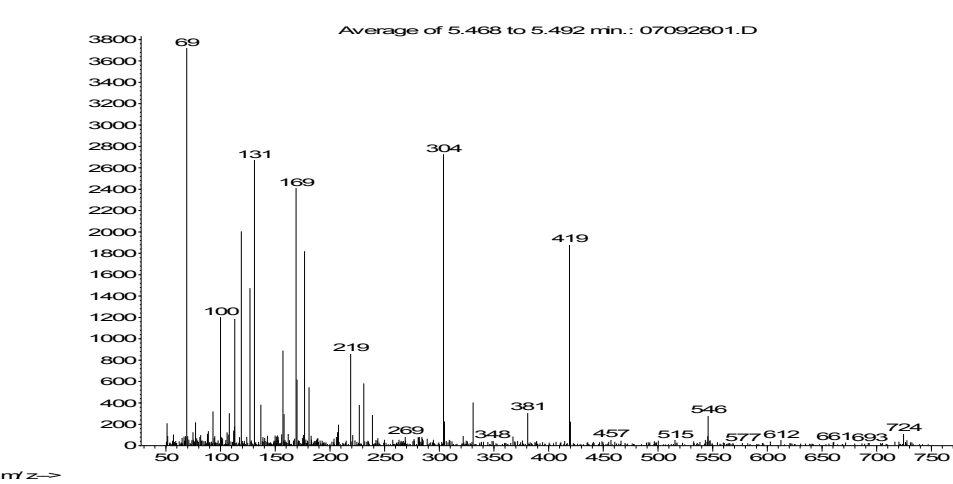
C2H5I 156, 127



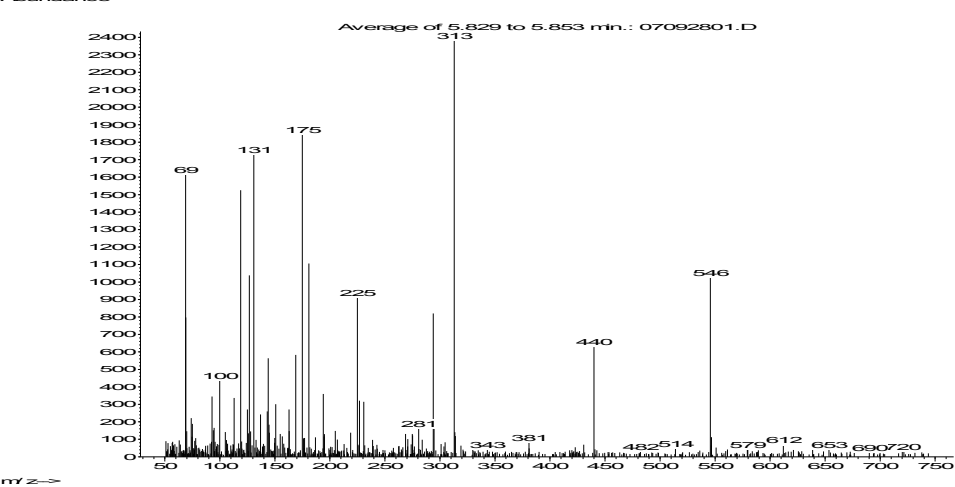
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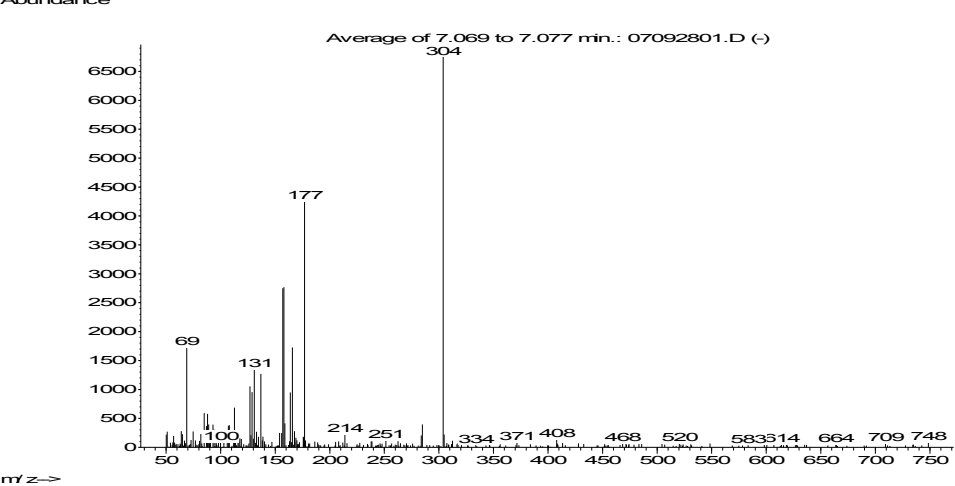
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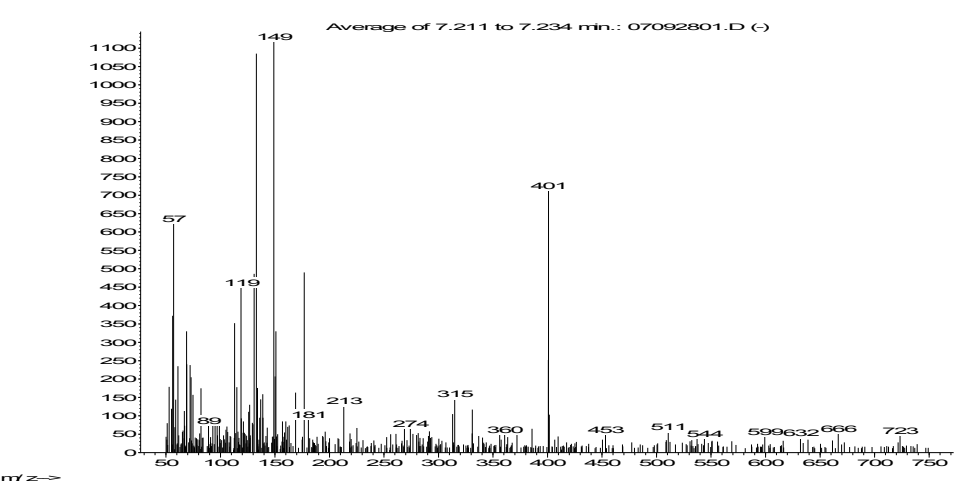
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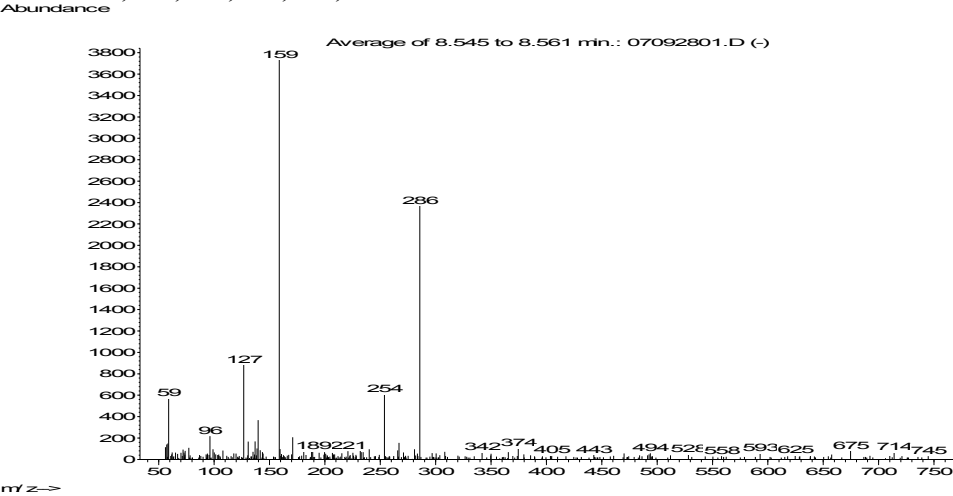
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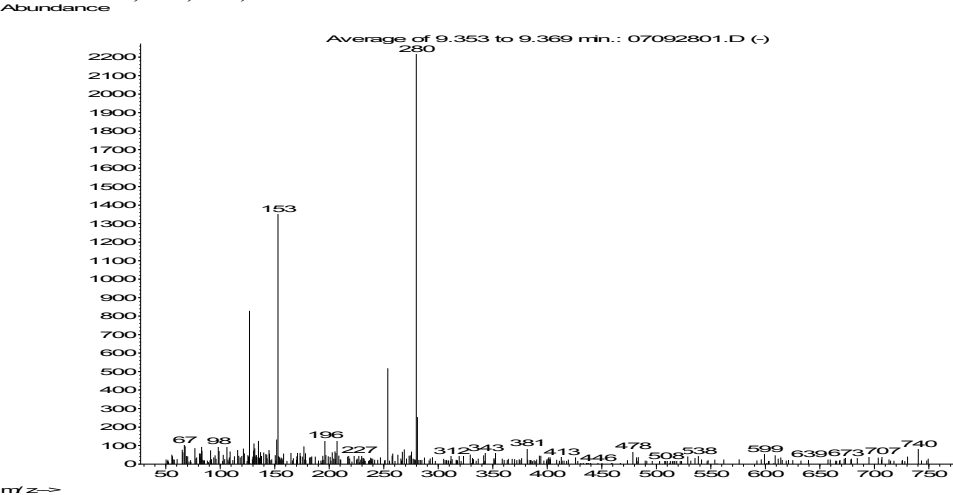
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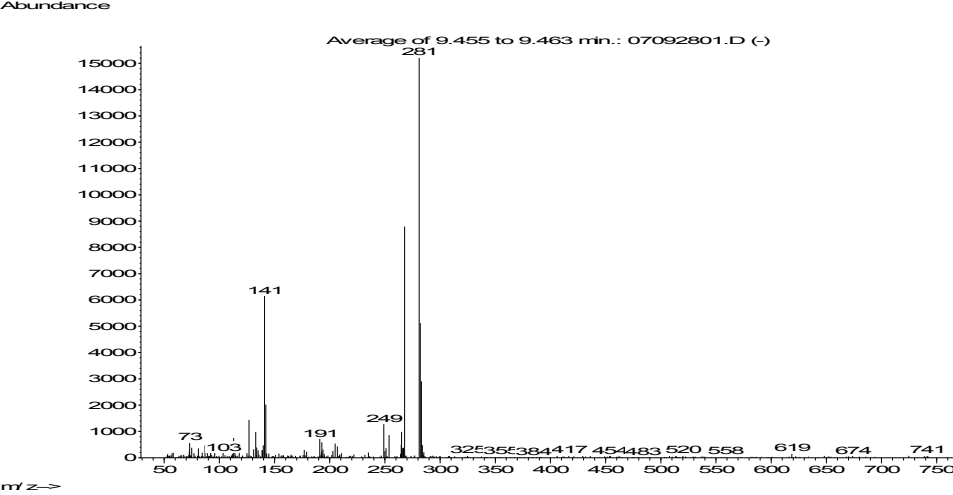
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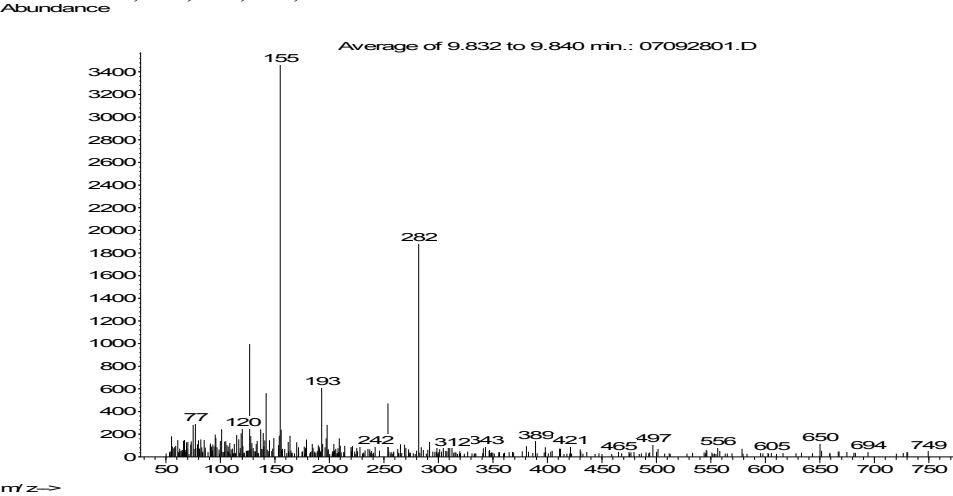
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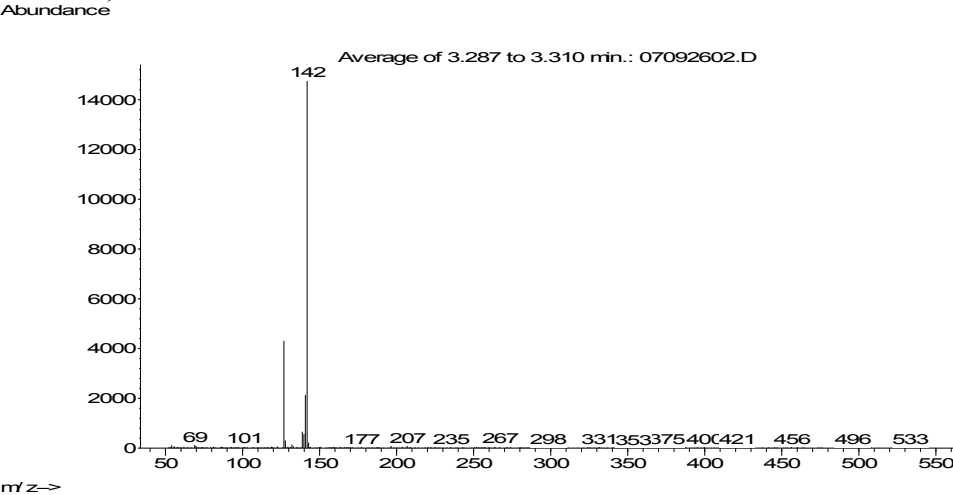
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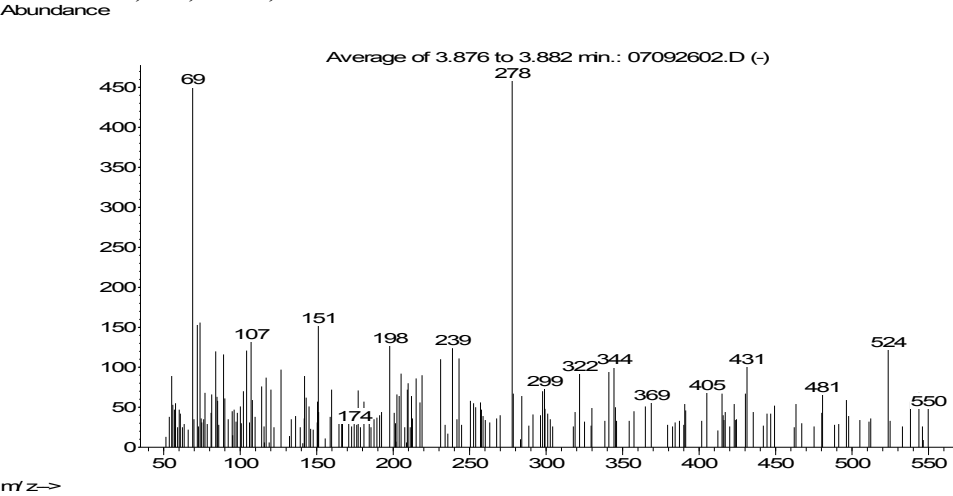
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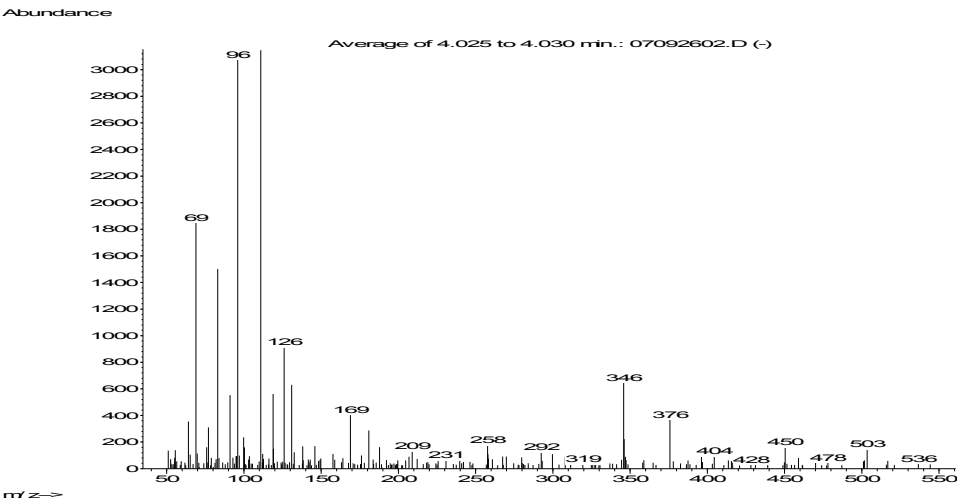
CH₃I 142, 127



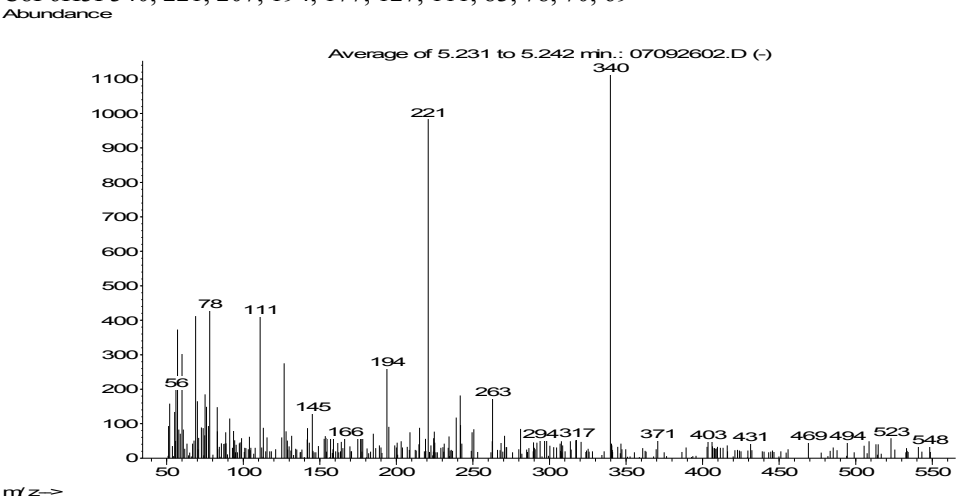
C₃F₆HI 278, 142, 127, 72, 69



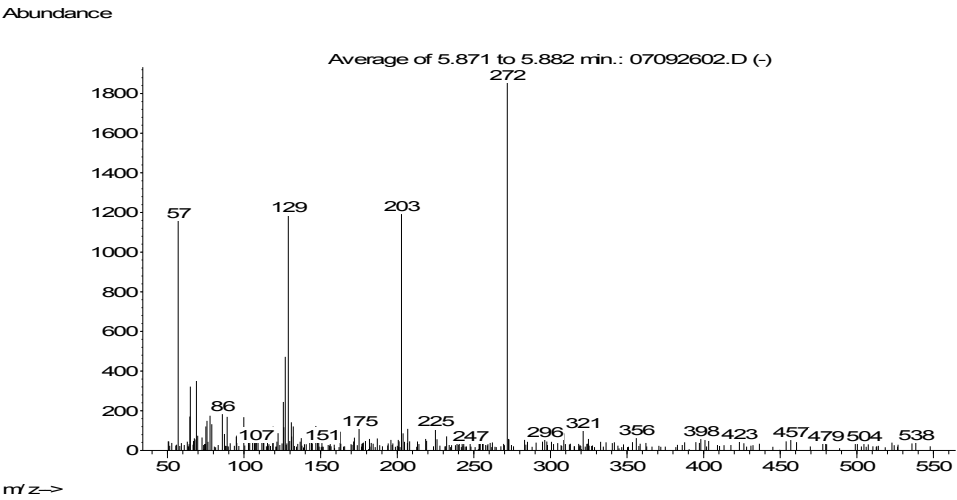
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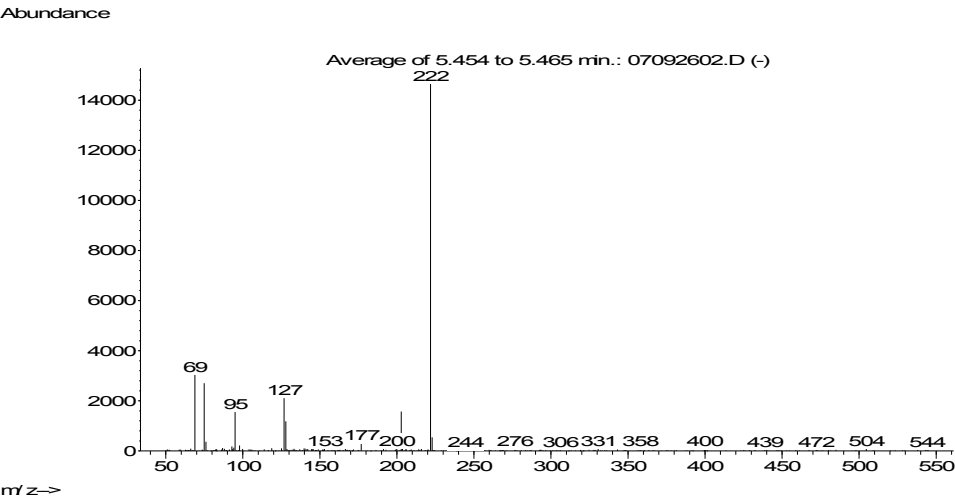
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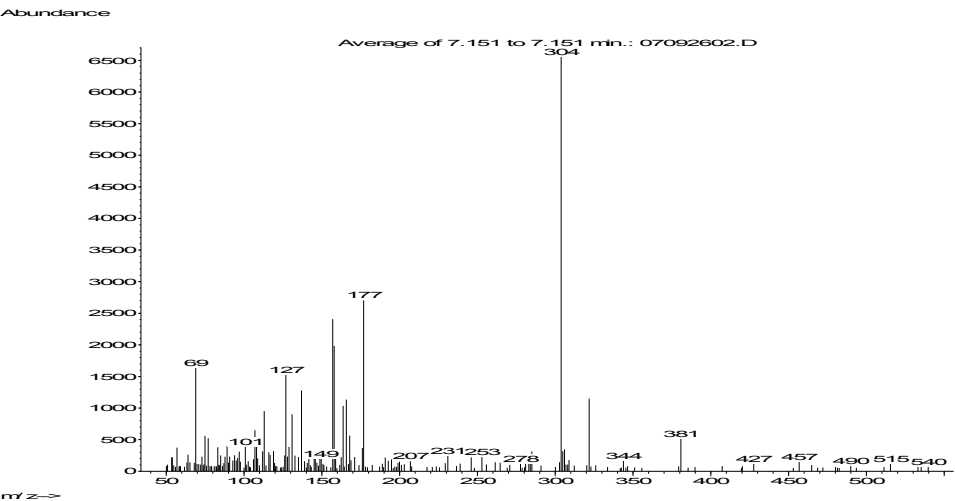
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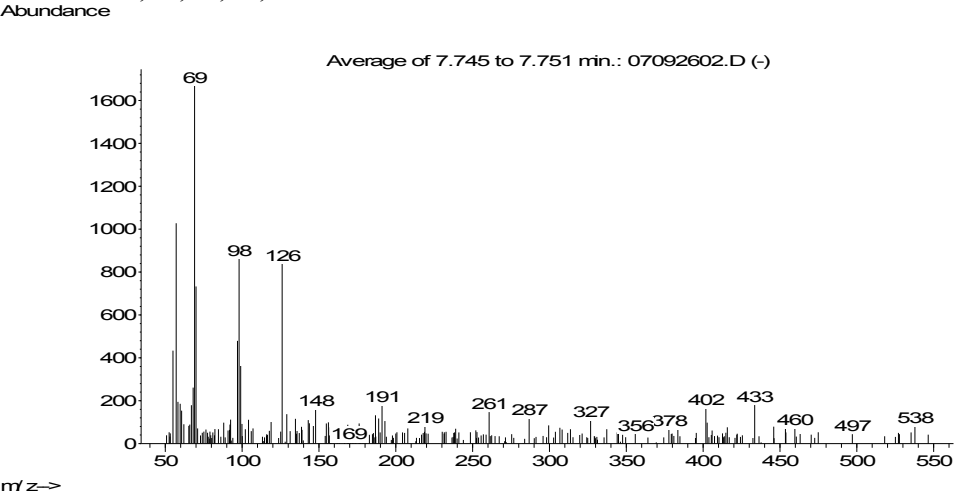
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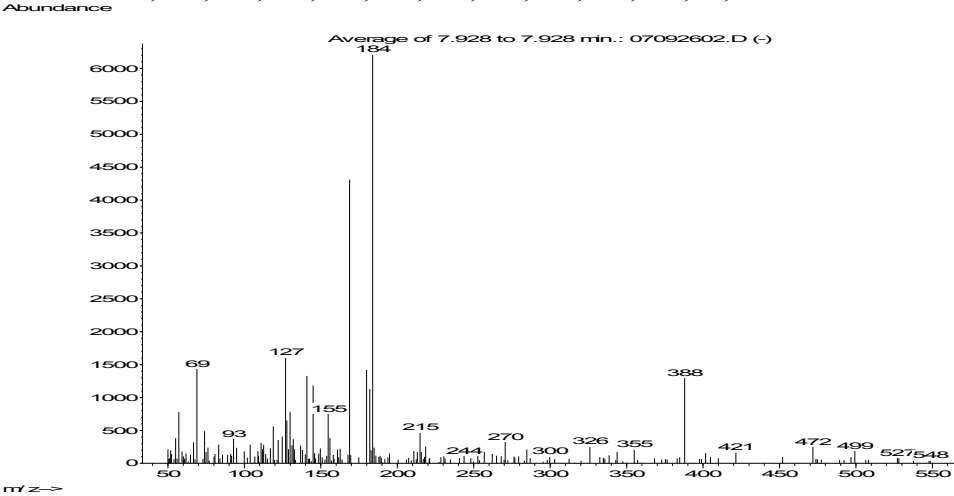
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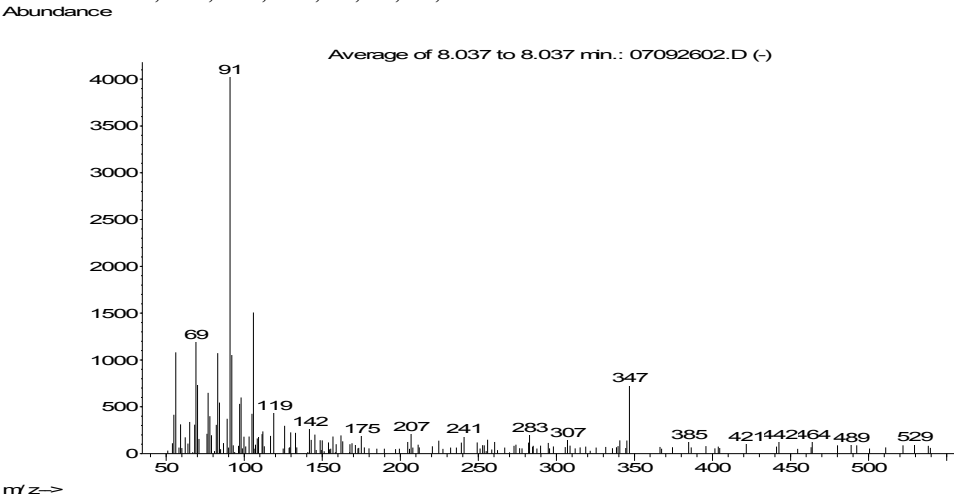
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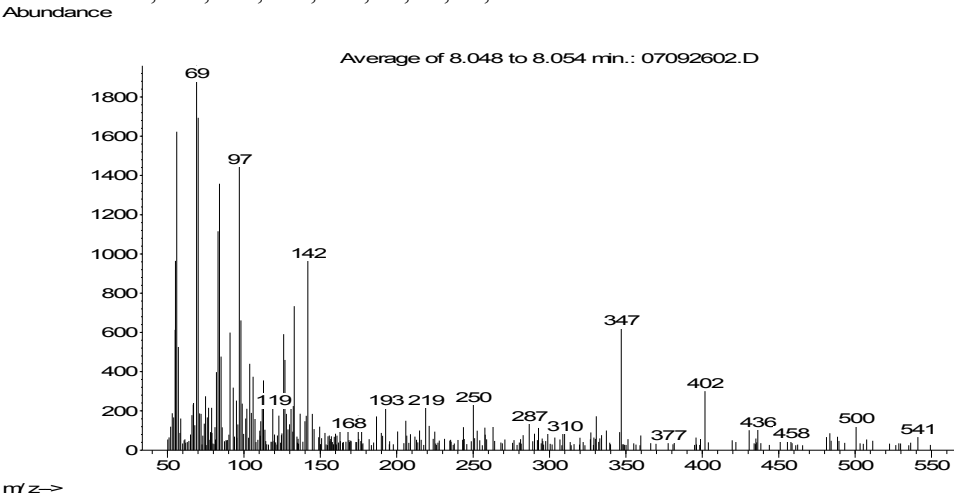
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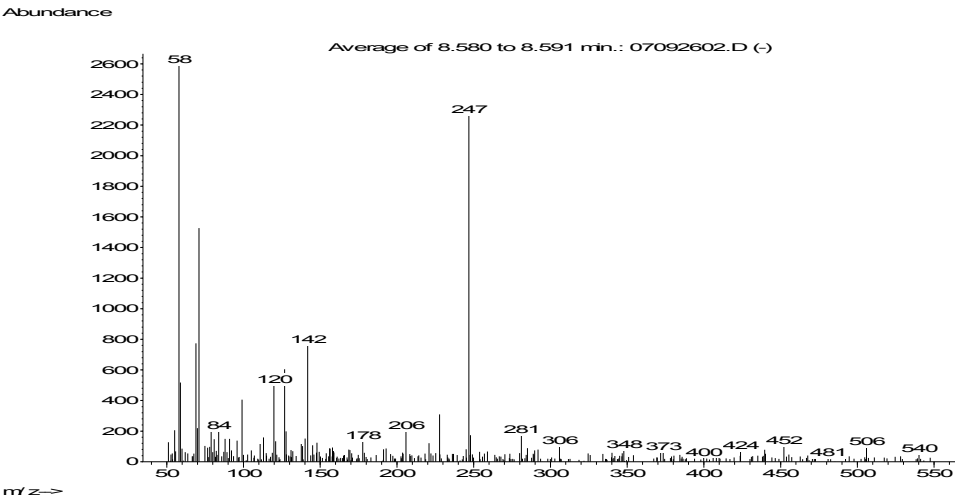
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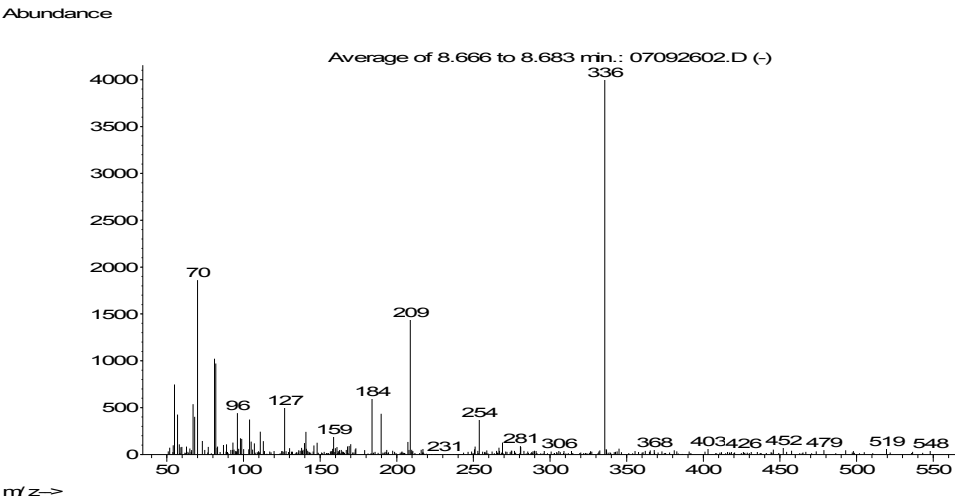
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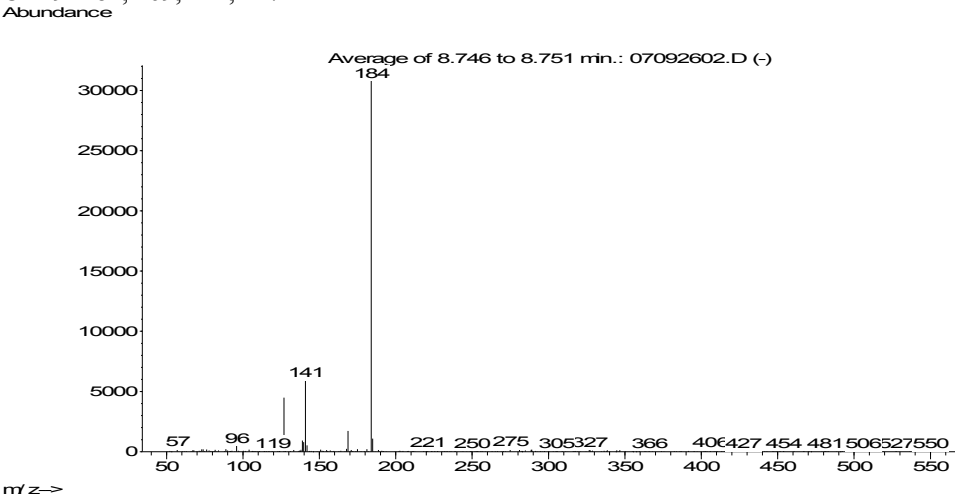
C5F3HI 247, 206, 142, 127, 114, 71, 69, 58



C₂F₃HI₂ vs. C₅FH₃I₂ 336, 254, 209, 184, 142, 127, 104, 81, 70, 67, 57, 55



C₄H₉I 184, 169, 141, 127



CH2I2 268, 254, 198, 169, 141, 127

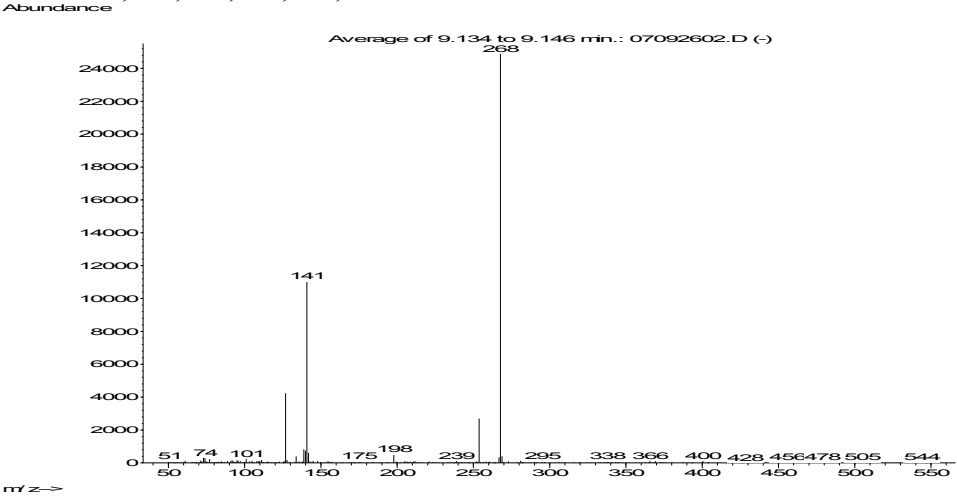
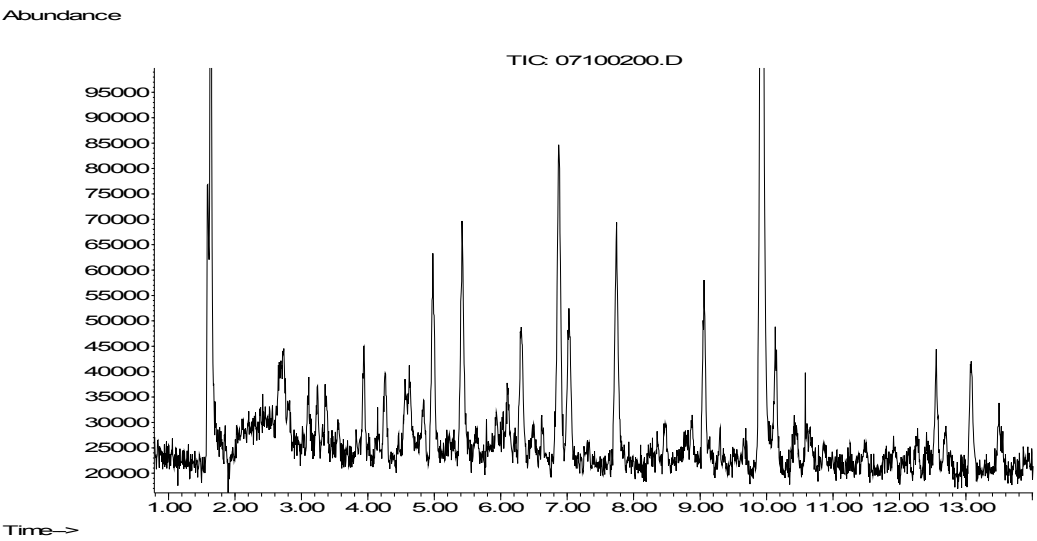
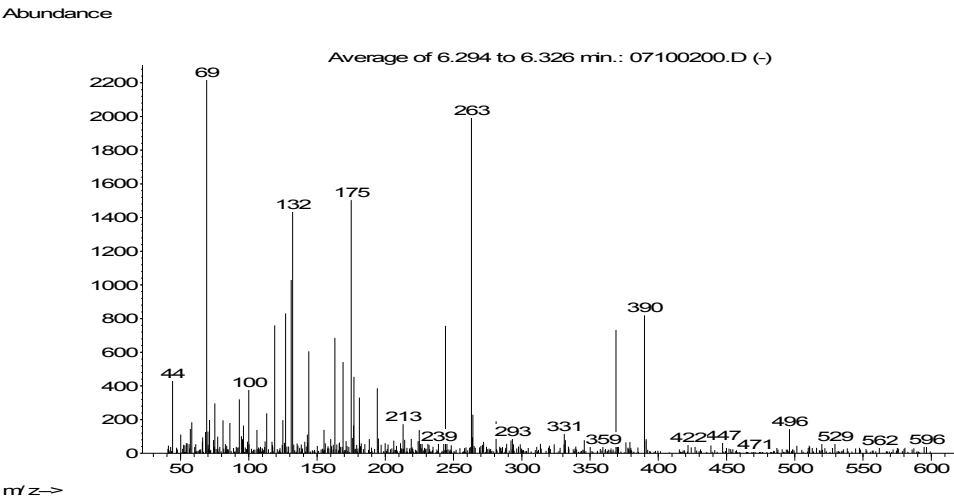


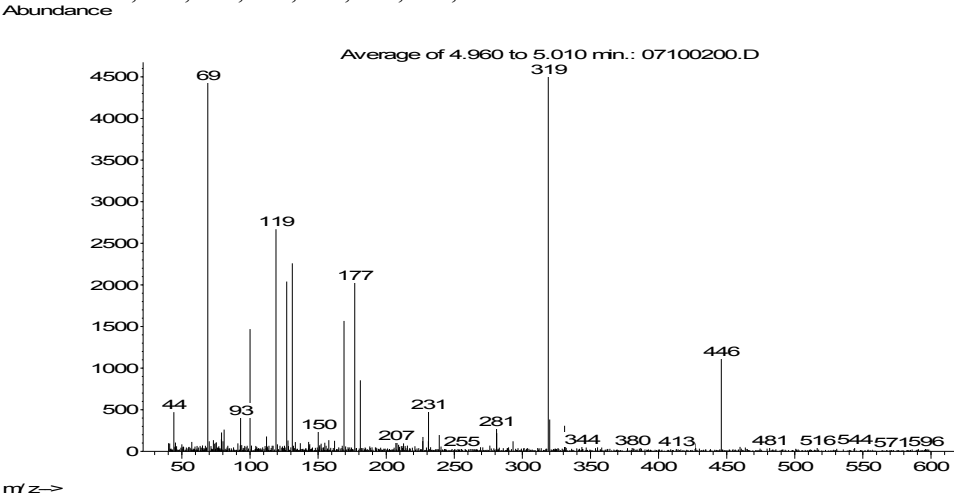
Fig. S8 (below). Total ion chromatogram (TIC) of gaseous intermediates trapped during the course of 2.5h-photolysis of aqueous PFOA with KI of 10 mM, and extracted ion chromatograms (EIC).



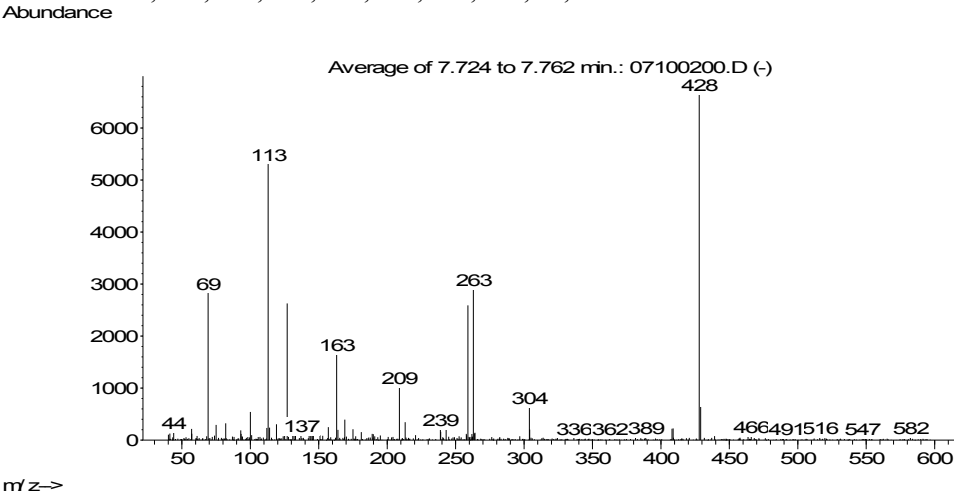
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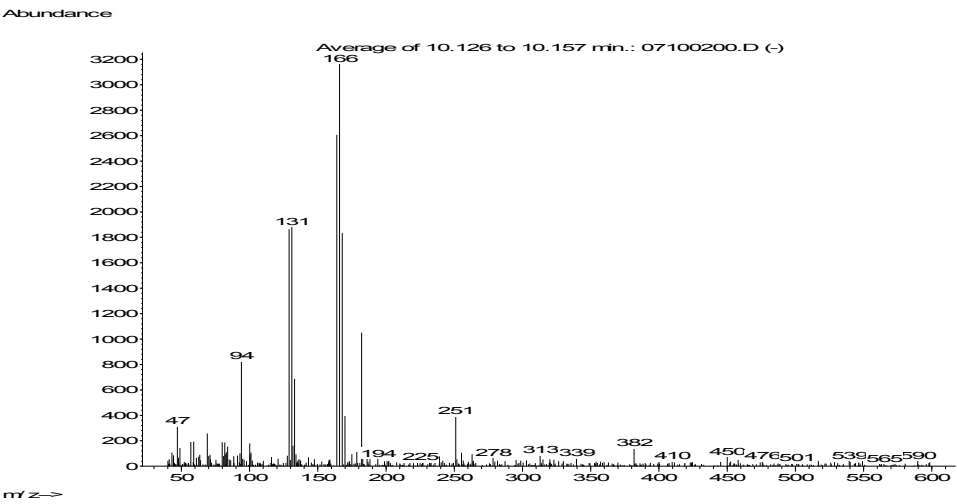
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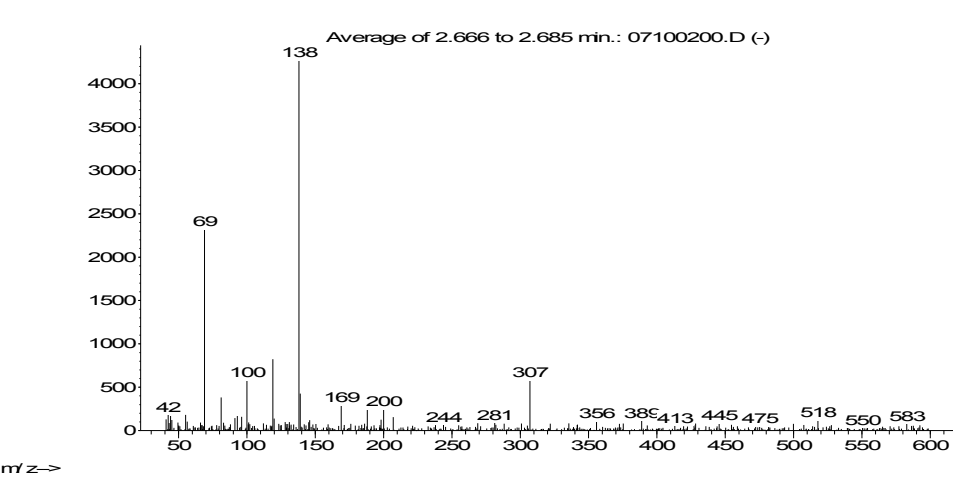
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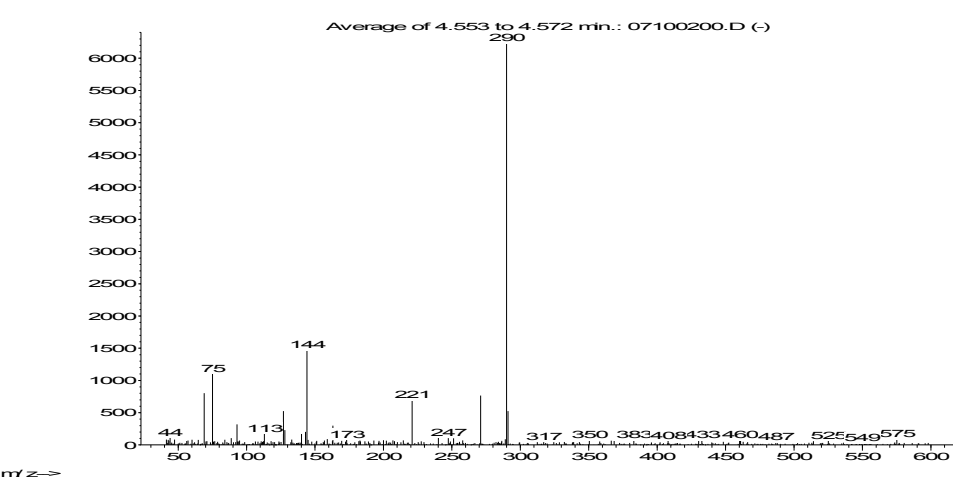
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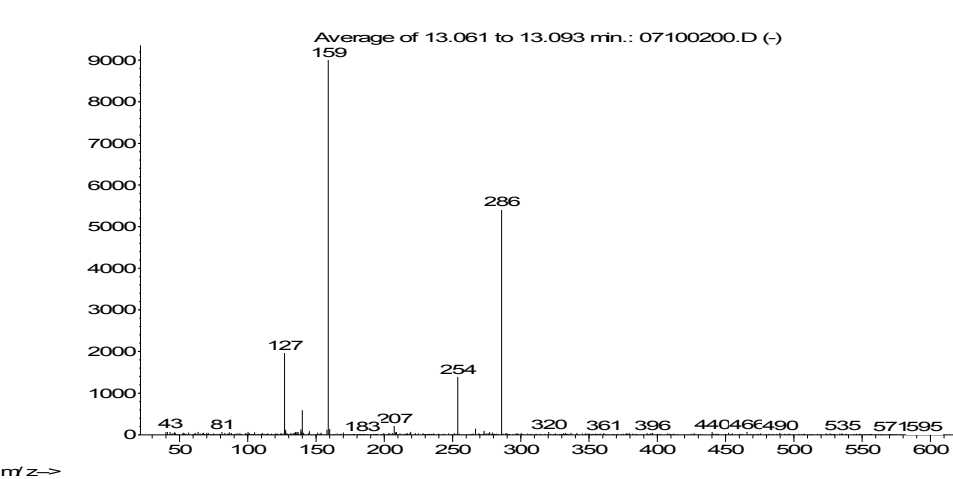
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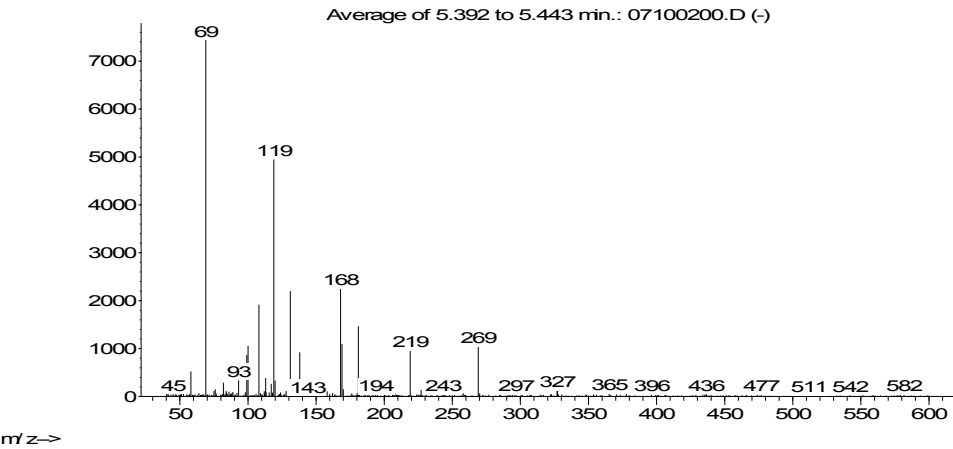
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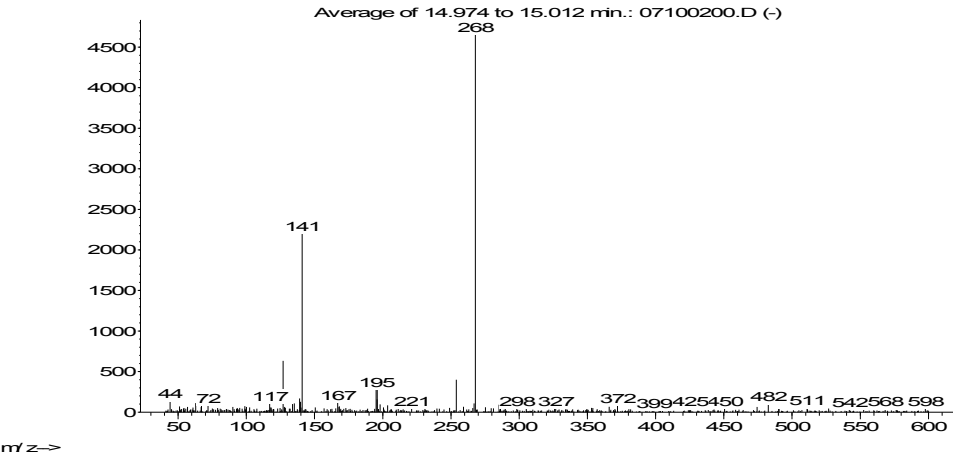
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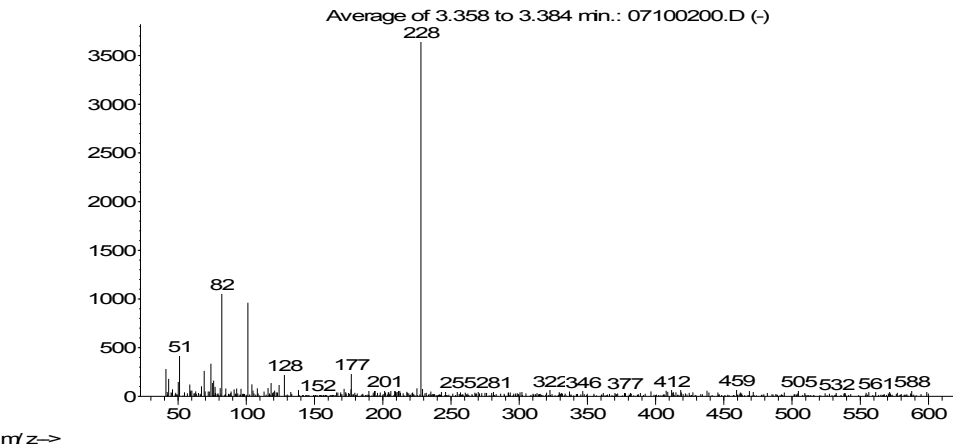
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Abundance



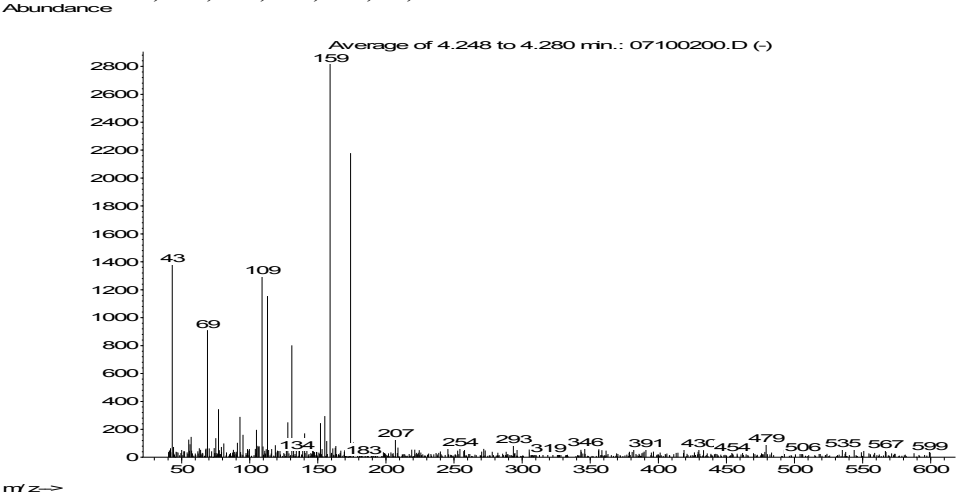
CH2I2 268, 254, 141, 127
Abundance



C2F4HI 228, 209, 127, 101, 82, 693
Abundance



C3F7H5? 174, 159, 131, 113, 109, 93, 69



CH3I 142

